

APPLICATION OF THE STEVENS REARRANGEMENT-ELIMINATION PROCEDURE
FOR THE SYNTHESIS OF [2.2]METAPARACYCLOPHANE-1,9-DIENE.⁽¹⁾

V. Boekelheide and Peter H. Anderson

Department of Chemistry, University of Oregon, Eugene, Oregon 97403

(Received in USA 20 January 1970; received in UK for publication 24 February 1970)

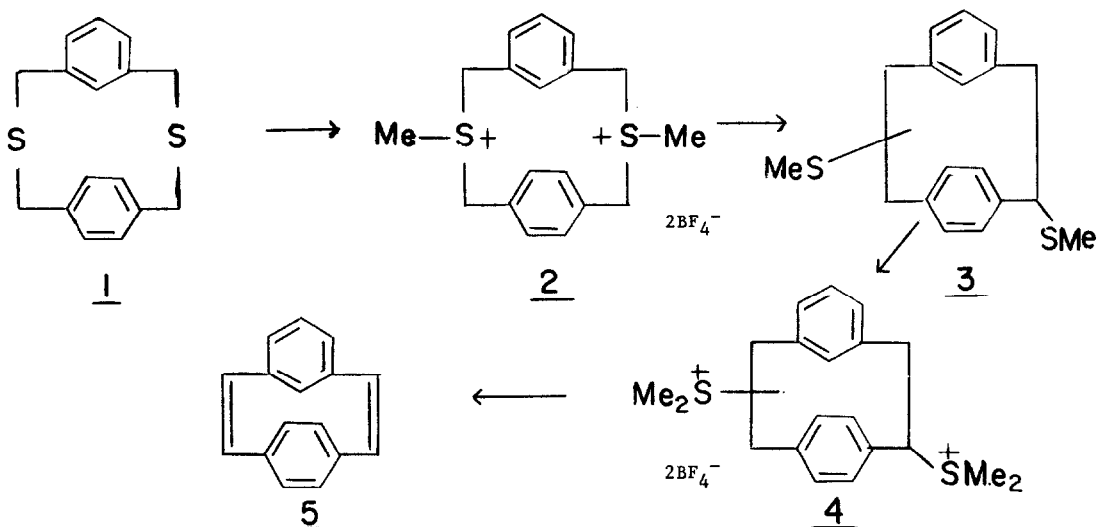
In an accompanying communication,⁽²⁾ we describe a procedure for replacing a sulfide linkage with a carbon-carbon double bond. To test the generality of this method we have investigated its use for the synthesis of [2.2] - metaparacyclophane-1,9-dienes, a class of compounds having some theoretical interest.⁽³⁾

The reaction of p-xylylene dibromide with m-xylylene dimercaptan in a basic solution of aqueous ethanol gave 2,11-dithia[3.3]metaparacyclophane (1) as white crystals, m.p. 158.5-159.5°, in 30% yield. The properties of 1 are in agreement with those reported earlier by Vögtle using an alternate route.⁽⁴⁾ The reaction of 1 with dimethoxycarbonium fluoroborate⁽⁵⁾ proceeded smoothly in essentially quantitative yield to give the disulfonium salt 2 as white crystals, m.p. 210° dec.^a The Stevens rearrangement of 2 using sodium hydride in dimethyl sulfoxide gave 3 as a colorless oil in 67% yield.

Even though 3 was a mixture of stereoisomers, its nmr spectrum clearly showed the typical [2.2]metaparacyclophane pattern⁽⁶⁾ with the internal proton at the 8-position appearing as a broad singlet at τ 4.70. Treatment of 3 with dimethoxycarbonium fluoroborate then gave 4 in 87% yield as white crystals, decomposing above 236°. No attempt was made to separate the mixture of isomers represented by 4 but, instead, it was treated with n-butyllithium at -78°.

^a For the new compounds being reported satisfactory analyses and mass spectra, where appropriate, have been obtained.

This gave the desired [2.2]metaparacyclophane-1,9-diene (5) in 50% yield. The properties of this product were identical in all respects to those reported earlier for 5.⁽³⁾



Although an x-ray crystallographic analysis shows the aromatic rings of 5 to be inclined at an angle of 41° from perpendicular,⁽⁷⁾ the nmr spectrum of 5 in solution shows the four para-bridged aromatic protons as a singlet at τ 3.19.⁽⁸⁾ The convenience of the present method of synthesis of 5 should allow a closer examination of its geometry in solution as well as the question of electronic transmission effects between rings in the case of appropriate derivatives of 5.

REFERENCES:

1. We thank the National Science Foundation for their support of this investigation.
2. R.H. Mitchell and V. Boekelheide, *Tetrahedron Letters*, Accompanying Communication.
3. T. Hylton and V. Boekelheide, *J. Am. Chem. Soc.*, 90, 6887 (1968).
4. F. vögtle, *Chem. Ber.*, 102, 3077 (1969).
5. R.F. Borch, *J. Org. Chem.*, 34, 627 (1969).
6. D.J. Cram, R.C. Helgeson, D. Lock, and L.A. Singer, *J. Am. Chem. Soc.*, 88, 1324 (1966).
7. A.W. Hanson, *Acta Crystallographic*, In Press.
8. T. Hylton, Ph.D. Thesis, University of Oregon, 1969.